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(54) Title: PROCESS FOR THE ENZYMATIC PREPARATION OF A BETA-LACTAM DERIVATIVE

(57) Abstract

Process for the preparation of a β -lactam derivative, wherein in an enzymatic reaction a β -lactam nucleus is coupled to D-phenylglycine amide, and the enzyme, the solid phenylglycine and the β -lactam derivative are separated out, wherein the mixture obtained after the enzymatic reaction, from which mixture at least the enzyme and the solid D-phenylglycine have been removed, is treated with a suitable aldehyde at a pH between 7.5 and 8.5, whereby the Schiff base of D-phenylglycine amide is formed, and the Schiff base of D-phenylglycine amide is separated out. The treatment with a suitable aldehyde preferably takes place before the β -lactam derivative is separated out.

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5 PROCESS FOR THE ENZYMATIC PREPARATION OF A BETA-LACTAM DERIVATIVES.

The invention relates to a process for the preparation of a β -lactam derivative, wherein in an enzymatic reaction a β -lactam nucleus is coupled to D-phenylglycine amide, and the enzyme, the solid phenylglycine and the β -lactam derivative are separated out.

The enzymatic coupling of a β -lactam nucleus and an acylation agent to form a β -lactam derivative is described in WO-A-9201061. In particular esters and amides of phenylglycine and p-hydroxyphenylglycine are described as acylation agents. The recovery and purification of the product is effected in a known way, for instance by crystallization. Although the product can be obtained with a good yield and high purity in this way, in practice it is difficult to separate from each other for the purpose of re-use the substances that remain behind in the reaction mixture, such as non-converted phenylglycine amide, non-converted β -lactam nucleus, formed phenylglycine in solution, β -lactam derivative remaining in solution after crystallization.

Since in the enzymatic coupling reaction phenylglycine amide is present in a large excess relative to the β -lactam nucleus - the optimum molar ratio of phenylglycine amide to β -lactam nucleus is often between 2 and 9 - in order to secure a high yield from the enzymatic reaction, it is notably important to recover the non-converted phenylglycine amide in order to obtain a commercially attractive process.

The invention now provides a process whereby more than 85% of the phenylglycine amide can be recovered

- 2 -

in pure form in a simple way, even before the β -lactam derivative is separated out.

This is achieved according to the invention in that the mixture obtained after the enzymatic reaction, from which mixture at least the enzyme and the solid phenylglycine have been removed, is treated with a suitable aldehyde at a pH between 7.5 and 8.5, whereby the Schiff base of phenylglycine amide is formed, and the Schiff base of phenylglycine amide is separated out.

For surprisingly it has been found that in this way the Schiff base of the phenylglycine amide is formed, while the aldehyde does not form Schiff bases with the other compounds (amines), which means that the other compounds remain dissolved in the mixture. This is all the more surprising as it is generally known that when amines are contacted with an aldehyde, they react to form a Schiff base and then precipitate as such. The β -lactams which are present in the reaction mixture appear to remain virtually completely in solution, however, while the phenylglycine amide appears to have virtually completely reacted with the aldehyde added.

It should also be pointed out that US-A-4172846 describes a process for the separation of L-phenylglycine from D-phenylglycine amide via formation of a Schiff base. That publication teaches nothing about the possible selective separation of phenylglycine amide from the mixture which also contains β -lactams. Moreover, in the examples the separation is in all cases carried out at a pH of about 10.5, which pH values are prohibitive in the present process on account of the instability of the β -lactam derivatives at such a high pH. The pH values in the process according to the invention are between 7.5 and 8.5. At such pH values, part of the phenylglycine amide will be present in the form of salt in the reaction mixture, as a result of which the formation of Schiff base is hindered.

The mixture that is obtained after the enzyme

and the solid phenylglycine have been removed from the reaction mixture obtained after the enzymatic coupling reaction, will in general contain the β -lactam derivative formed, D-phenylglycine amide, remaining β -lactam nucleus and D-phenylglycine formed in quantities, relative to the β -lactam derivative formed, of for instance 1-7 eq. of D-phenylglycine amide, 0.02-1 eq. of β -lactam nucleus, 0.1-2 eq. of D-phenylglycine. The invention also relates to the removal of phenylglycine amide from such a mixture.

10 The process according to the invention also offers the advantage that D-phenylglycine amide can be separated out in a simple manner, for instance by filtration or extraction, and in a pure form. Preferably, the D-phenylglycine amide is separated out before the β -lactam derivative is recovered via crystallization. For this offers the additional advantage that the recovery of the β -lactam derivative as a product is simplified, since the solubility of the β -lactam derivative is higher when the D-phenylglycine amide is still present in the mixture.

15 The β -lactam derivative can be recovered and purified in a known manner.

 Within the framework of the invention, by β -lactam derivative is understood semisynthetic β -lactam obtained by acylation of a β -lactam nucleus with an activated form of D-phenylglycine as acylation agent. They are mostly used as antibiotics, such as Ampicilline, Cefalexine and Cefachlor. Known β -lactam nuclei are for instance 6-aminopenicillanic acid (6-APA), 7-aminocephalosporanic acid (7-ACA), 7-amino-3-chloro-3-cephem-4-carboxylate, 7-aminodesacetoxy-cephalosporanic acid (7-ACDA), etc.

 The enzymatic coupling reaction can be carried out in a known manner, for instance as described in WO-A-92/01061.

35 The enzyme used in the process according to the invention can be any suitable enzyme which catalyzes the reaction. The enzyme may for be derived from known

microorganisms such as Acetobacter, Aeromonas,
Aphanocladium, Athrobacter, Bacillus Cephalosporium,
Corynebacterium, Escherichia, Flavobacterium, Pseudomonas
and Xanthomonas.

5 The D-phenylglycine amide that is used in the
enzymatic reaction can be applied as free D-phenylglycine
amide or as salt of an acid, for instance an acid with a
 $pK_a < 5$, such as acetic acid, formic acid, sulphuric acid,
hydrochloric acid or nitric acid.

10 The concentration of reactants is not critical
and is preferably chosen such that upon completion of the
enzymatic reaction all components, except for the D-
phenylglycine formed, are still just in solution. Water
can be used as solvent, but the use of an organic solvent
15 is also possible. The temperature at which the enzymatic
reaction is carried out is mostly between 0 and 40°C; the
pH is mostly between 5 and 8.5.

 Preferably, benzaldehyde is used in the
formation of the Schiff base of D-phenylglycine amide.

20 Benzaldehyde is advantageous in that the
separation of the Schiff base and the recovery of
benzaldehyde are simple. Another advantage of benzaldehyde
is that it is immiscible with water, so that as extraction
agent it is preferred by far over other extraction agents,
25 because the resulting Schiff base of the optically active
phenylglycine amide dissolves in the benzaldehyde and the
other components of the reaction mixture in the water
phase.

 The process according to the invention can also
30 be carried out with other aldehydes, provided they meet
the following requirements:

1. they should readily form water-immiscible Schiff bases
of the D-phenylglycine amide;
2. the Schiff bases formed should readily decompose
35 without decomposition of the aldehyde;
3. there should be a clear difference in the solubility in
water and in organic solvents between the Schiff bases of

the D-phenylglycine amide and the other components of the reaction mixture.

Benzaldehyde and substituted benzaldehydes are excellent examples of compounds that meet all these requirements very well.

By benzaldehyde are also understood substituted benzaldehydes, such as lower alkyl benzaldehyde, halogen benzaldehyde, nitrobenzyl aldehyde and lower alkoxybenzyl aldehyde. Lower alkyl or lower alkoxy means alkyl or alkoxy with 1-5 C-atoms.

The reaction with benzaldehyde with formation of Schiff base can be carried out at temperatures of 0-50°C, preferably 5-45°C. If for the formation of the Schiff base equimolar quantities of benzaldehyde, for instance 9-2, in particular 0.95-1.1 equivalents relative to the phenylglycine amide are applied without another solvent for the Schiff base of the amide, then a precipitate of the Schiff base of D-phenylglycine amide is obtained. The other components remain dissolved in the mother liquor. If an excess of benzaldehyde is used, the benzaldehyde will not only act as reaction agent but also as solvent, and two layers will be obtained. It is also possible to apply mixtures of benzaldehyde and other solvents, such as mixtures with toluene, chloroform, methylisobutyl ketone, tetrachloroethene, ethyl acetate and butyl acetate.

Recovery of the D-phenylglycine amide from the corresponding Schiff base can be simply effected through acidification with an equimolar quantity of strong acid, e.e. sulphuric acid (up to pH = 2-3), which results in decomposition of the Schiff base into the aldehyde and D-phenylglycine amide. The D-phenylglycine amide thus obtained can be supplied back to the enzymatic coupling step.

The invention will now be further elucidated by means of the following example, without being restricted thereto.

Example

- To a 500 ml flask were successively supplied (corresponding to a reaction mixture obtained after enzymatic preparation of Cefalexine, followed by removal of the enzyme and solid D-phenylglycine): 0.56 g of 7-ADCA, 15.1 g of cefalexine, 1 H₂O, 1.18 g of D-phenylglycine, 19.9 g of D-phenylglycine amide, 1/2 H₂SO₄, 185.4 g of H₂O, 8.1 g of H₂SO₄ (98%) and 19.8 g of 25% ammonia solution.
- 10 The pH of this solution (250.0 g) was 8.0.
- At 20°C, 1 eq. (10.6 g) of benzaldehyde was added in 5 minutes with stirring. After 2 hours the resulting white solid substance was filtered off, washed with 3x20 ml H₂O and dried.
- 15 Yield of D-phenylglycine amide Schiff base: 20.8 g (> 99% pure); amounts of Cefalexine, phenylglycine and 7-ADCA each less than 0.5%.

C L A I M S

1. Process for the preparation of a β -lactam derivative,
wherein in an enzymatic reaction a β -lactam nucleus
5 is coupled to D-phenylglycine amide, and the enzyme,
the solid D-phenylglycine and the β -lactam derivative
are separated out, characterized in that the mixture
obtained after the enzymatic reaction, from which
mixture at least the enzyme and the solid
10 D-phenylglycine have been removed, is treated with a
suitable aldehyde at a pH between 7.5 and 8.5, and
the Schiff base of D-phenylglycine amide is separated
out.
2. Process according to claim 1, wherein the treatment
15 with a suitable aldehyde takes place before the
 β -lactam derivative is separated out.
3. Process for the recovery of D-phenylglycine amide
from a mixture containing a β -lactam derivative, D-
phenylglycine amide, D-phenylglycine and a β -lactam
20 nucleus, characterized in that the mixture is treated
with a suitable aldehyde, whereby the Schiff base of
D-phenylglycine amide is formed, and the Schiff base
of D-phenylglycine amide is separated out.
4. Process according to claim 3, characterized in that
25 the mixture contains, relative to the amount of
 β -lactam derivative, 1-7 equivalents of D-
phenylglycine amide, 0.1-2 equivalents of D-
phenylglycine and 0.02-1 equivalent of β -lactam
nucleus.
- 30 5. Process according to any one of the claims 1-4,
wherein the β -lactam nucleus is chosen from the group
comprising 6-APA, 7-ACA, 7-amino-3-chloro-3-cephem-4-
carboxylate and 7-ADCA.
6. Process according to any one of the claims 1-5,
35 wherein benzaldehyde is used as aldehyde.

7. Process according to any one of the claims 1-6, wherein the treatment with the aldehyde takes place at a temperature of 5-45°C.
8. Process according to any one of the claims 1-7,
5 wherein 0.95-1.1 equivalents of benzaldehyde is applied, relative to the amount of D-phenylglycine amide.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/NL 94/00161

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 C12P35/04 C12P37/04 C12P17/10 C07C231/20

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 C12P C07C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
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Y	WO,A,92 01061 (NOVO NORDISK) 23 January 1992 cited in the application see page 7 - page 8; claims ---	1
Y	WO,A,92 12782 (NOVO NORDISK) 6 August 1992 see claims ---	1
Y	EP,A,0 442 585 (STAMICARBON) 21 August 1991 see the whole document ---	1
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☒ Patent family members are listed in annex.

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C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

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INTERNATIONAL SEARCH REPORT

Information on patent family members

Internat Application No

PCT/NL 94/00161

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